Chemistry 1142 Syllabus General Chemistry 2 Spring 2025

Instructor:	Dr. Andrew Napper
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Office hours:	W 10:00 – noon R 3:30 – 4:30 P.M. F 11:00 – noon
Lecture:	TR 9:30 – 10:50 A.M. (MAS 206)
Quiz:	A 10-minute quiz will be given at the start of Tuesday's lecture. It will cover material from the previous week's lectures.
Lab:	T 11:00 - 1:50 P.M. (Massie 339) Section 01 R 11:00 - 1:50 P.M. (Massie 339) Section 02
Attendance policy:	Attendance at laboratories is required. Two or more unexcused lab absences will result in a grade of F for CHEM1142. If you are more than 15 minutes late to lab, this will count as an absence.
Excused absence policy:	In case of illness, accident, family emergency, or university-sponsored activity, you may be excused from labs, quizzes, and/or homework. In case of a missed exam, a make-up exam will be provided. For university-sponsored activities, you must email me at least one-week before your absence and let me know what quizzes/labs/exams you will be missing. Failure to do so may result in a grade of zero for any missed assignment. If you feel ill on the day of an exam or an in-class activity please report your illness to me immediately. Failure to do so within 24 hours will result in a grade of zero. If you test positive for COVID-19 or are placed in quarantine, you are responsible for notifying me of the dates you will be absent and making up any missed coursework For other absences, suitable documentation (such as a doctor's note, police accident report, etc.) must be provided by email within one-week of the excused absence . For absences longer than one week, an academic dean or the dean of students may issue you an excused absence which you can present to your instructor. Unexcused absences will result in a grade of zero for the assignment.
Grading:	3 exams

Final exam:	Thu, May 01 @ 8 AM in Massie 206
Final exam information:	The final exam is an American Chemical Society standardized final. It is fully comprehensive, covering material from CHEM 1141 and 1142.
Required materials:	Chemistry, 15/e (eBook, accessible via BlackBoard) —Jason Overby Note: you can also purchase a loose-leaf copy of the book for around \$30 from the bookstore: ISBN: 9781266334474 Aleks360 (online homework) —Bundled with the textbook or a separate access card Chemistry 1142 Lab Manual, Spring 2025 —Andy Napper A non-programmable scientific calculator (TI-30XIIS) Safety goggles or visorgogs (ANSI Z-87 approved)
	Salut Rassan an analogs (MISI 2-0/ approved)

Grading scale:

e				
	% Grade	% Grade	%	Grade
	>93 A	77–80 C+	60–63	D-
	90–93 A–	73–77 C	<60	F
	87–90 B+	70–73 C–		
	83–87 B	67–70 D+		
	80-83 B-	63–67 D		
Blackboard course-site:	Notes, handouts, an	nd other useful piec	es of info	mation will be available at the
	following URLs: http://blackboard.s http://chem1142.ss			
Online homework:	basis. The homework problems. Each hom be due the followin • You will be able to + Homework" tab or	k set may consist of t nework set will be m g Tuesday by 9 A.M. access online assignn n the left side of the s ss course-fee provide:	utorials, ho nade availa nents on Bl ccreen. s you with	ework will be assigned on a weekly omework problems, and review a ble on Friday by 5 P.M. and will lackboard. They are in the "eBook access to the textbook and an
In-class/online activities:	In-class activities worth various points will be completed, unannounced, during the semester. You must be present for the ENTIRE class period to receive credit for in class activities. No make ups will be allowed; however, the excused absence policy will apply. For some in class activities you may be required to watch an online lecture before coming to class. There may also be some online assignments, which will be completed outside of class.			
Cell-phone policy:	-	oooks, iPads, Surfac	es, smart v	, such as laptop computers, watches, etc.) are not permitted
General education program:	Chemistry 1142 cou Education Program			nce component of the General c <i>Reasoning</i> .

Study requirements:	To be successful in General Chemistry, you will need to study <i>at least</i> two hours outside of the classroom, for every hour spent in <i>lecture</i> .
Lecture material: <i>Chapter 11</i> <i>Chapter 12</i> <i>Chapter 13</i> <i>Chapter 14</i> <i>Chapter 15</i> <i>Chapter 16</i> <i>Chapter 17</i> <i>Chapter 18</i> <i>Chapter 19</i>	We will be covering the following chapters in your textbook: Intermolecular Forces and Liquids and Solids Physical Properties of Solutions Chemical Kinetics Exam 1 (5 PM, Feb 27, 2025 – MAS020) Chemical Equilibrium Acids and Bases Acid-Base Equilibrium and Solubility Equilibria Exam 2 (5 PM, Apr 10, 2025 – MAS020) Entropy, Gibbs Energy, and Equilibrium Electrochemistry Nuclear Chemistry Exam 3 (5 PM, Apr 24, 2025 – MAS020)
Exams 1 – 3:	Exams 1 – 3 will be held in Massie 020 from 5 P.M. – 7 P.M. A make-up exam will be available for students who have a time conflict or an official University excused absence.
Homework problems:	Problem solving is an <i>essential</i> part of your study of chemistry. As you study, you should be working problems from your textbook on each topic. In addition, it is strongly recommended that you work all of the problems from the Self-Assessment Quizzes at the end of each chapter.
University ADA statement: Religious accommodations:	https://www.shawnee.edu/syllabus-statements
Grading errors:	If you notice a grade error on BlackBoard for quizzes, exams, etc.—you need to bring it to the instructor's attention in writing within one week of the due date (for an online assignment) or one week from the assignment being handed back (lab/exam assignments).

Order of labs:

Week Beginning	Tuesday	Thursday
January 13th	I	I
January 20th	2	2
January 27th	3	3
February 3rd	4	4
February 10th	5	5
February 17th	6	6
February 24th	7	7
March 3rd	Spring Brea	uk (No Lab)
March 10th	8	8
March 17th	9	9
March 24th	10	10
March 31st	II	II
April 7th	12	12
April 14th	13	13
April 21st	14	14

Laboratories:

1. Check-in, safety, and review worksheet

- 2. Intermolecular forces
- 3. Nine-bottles—An adventure in chemical identification
- 4. Colligative properties: freezing point depression
- 5. Kinetics dry lab
- 6. Determining a rate law using spectrophotometry
- 7 Spectrophotometric determination of aspirin content in commercial tablets
- 8. Determining an equilibrium constant using spectrophotometry
- 9. pH of acid solutions and salt solutions
- 10. pHun with buffers!
- 11. Determining K_{sp} for lead(II) iodide
- 12. Thermodynamics of KNO3 dissolving in water
- 13. Electrochemical cells
- 14. Check-out

Laboratory information: Safety goggles or visorgogs are required to be worn for all laboratories. They must meet ANSI Z87 requirements (normally this information is permanently stamped on the goggles). Laboratory coats are recommended, but not required. Full length pants or full-length skirts are required to be worn in lab. Shoes that cover all parts of your feet are also required. If you are improperly dressed for lab, you will be asked to leave and awarded a zero for the lab assignment.

Lab reports must be turned in *at the start* of lecture, one week from the date of completion (or the next lecture period in case of holiday/closing/etc.). Late lab reports can be turned in at the start of the next lab period but will be subject to a two-point deduction. Turned in lab reports must have your full name clearly written on the front page to receive a grade. If reports are completed with your assigned lab partner, all names must be clearly written on the front page.

Who should take this course?The typical audience for this course is: science, engineering, pre-pharmacy, pre-
medicine, and science education majors. You may also be taking this course if you are
interested in chemistry (yay!), are seeking to satisfy the natural sciences general
education category, or curious about how things work.

Is chemistry hard?	Yes. But not impossible. Consider setting aside several hours a week to practice end-of- chapter homework problems, forming a study group, re-reading your Aleks360 assignments, reading the textbook, and quizzing yourself. Reviewing old material every few weeks has been shown to dramatically improve retention of material in college!
What should I do if I need help?	 If you need help—don't wait too long before you seek it out! The following is a partial list of options that are available to you: Student success center (SSC) tutoring. Stop by the SSC and sign up for a <i>free</i> tutor! Browse my course website for chapter objectives, old exams, lec. notes, quizzes, etc. YouTube. Amazing selection of videos on any topic you can think about. The <i>Khan Academy</i> videos are an excellent place to start. Office hours. I hold four office hours a week over three days. Stop by if you have any questions about the course!
How to study for this class	Buy a composition notebook to work problems in.
·	Skim through the textbook section before you come to each class
	□ After each class, but before the next class, go through the Example
0	problems in the chapter. Do the "Practice Exercise" problems after each example.
(m)+	You can click the "Answer" button in the eBook to reveal the solutions.
(□ In a separate notebook, answer the problems at the end of each chapter that go over the relevant sections. You can click the "Answer" button in the eBook to reveal the solutions
Oction	Before the exams:
Cation	
Pronunciation: [kat-ahy-uh n, -on]	□ One week before each exam, thoroughly read your notes, being sure to work out any
-noun, <i>Chemistry</i> 1. An ion with a paws-itive charge.	problems yourself that we went over. Try covering up my worked answers with a blank piece of paper and then working them yourself.
2. The cutest ion ever.	
	\Box Re-work the end-of-chapter and in-chapter problems
	□ Print off a practice exam and take it in a timed fashion. Print off the answers and

then grade yourself.

Hint: 90 % of your studying should consist of working problems!

End-of-chapter problems	It is strongly suggested that you work through the following problems to prepare for exams. The answers are given in the eBook for the even -numbered questions.
Chapter 11	4, 12, 16, 52, 64, 76, 82, 92, 104, 114, 138, 150
Chapter 12	4, 10, 14, 16, 22, 28, 34, 38, 42, 58, 60, 72, 78, 90, 106
Chapter 13	6, 14, 16, 24, 28, 32, 38, 40, 46, 49, 56, 60, 70, 74, 78, 82, 88, 100
Chapter 14	2, 8, 10, 16, 20, 26, 32, 40, 44, 54, 58, 64, 72
Chapter 15	4, 6, 14, 18, 26, 36, 42, 44, 50, 56, 64, 76, 80, 94, 102
Chapter 16	4, 6, 10, 20, 26, 30, 36, 50, 54, 66, 72, 76
Chapter 17	2, 4, 10, 14, 18, 24, 26, 36, 44, 52, 74
Chapter 18	2, 4, 6, 16, 20, 26, 30, 36, 48, 56
Chapter 19	6, 10, 20, 26, 34, 38
Disclaimer:	All dates and policies are subject to change as announced in class.