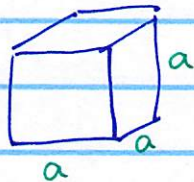


1/30/2019

Interesting application:

- imagine we have a "tiny crystal" of a special form of iron whose unit cell is fcc w/ edge length of 544 pm. (x-ray diffraction) **face-centered cubic**

What is its density? $d = \frac{m}{V}$ typically g
cm³ or mL



$$V = a^3$$

m = mass of Fe atoms in unit cell.

$$V = \left(544 \text{ pm} \times \frac{10^{-12} \text{ m}}{\text{pm}} \times \frac{\text{cm}}{10^{-2} \text{ m}} \right)^3 = \left(544 \times 10^{-10} \text{ cm} \right)^3 = 1.6099 \times 10^{-22} \text{ cm}^3$$

fcc: 8 atoms @ corner, 6 atoms @ faces.

$$\# \text{ atoms} = 8 \times \left(\frac{1}{8} \right) + 6 \times \left(\frac{1}{2} \right) = 4 \text{ atoms of Fe}$$

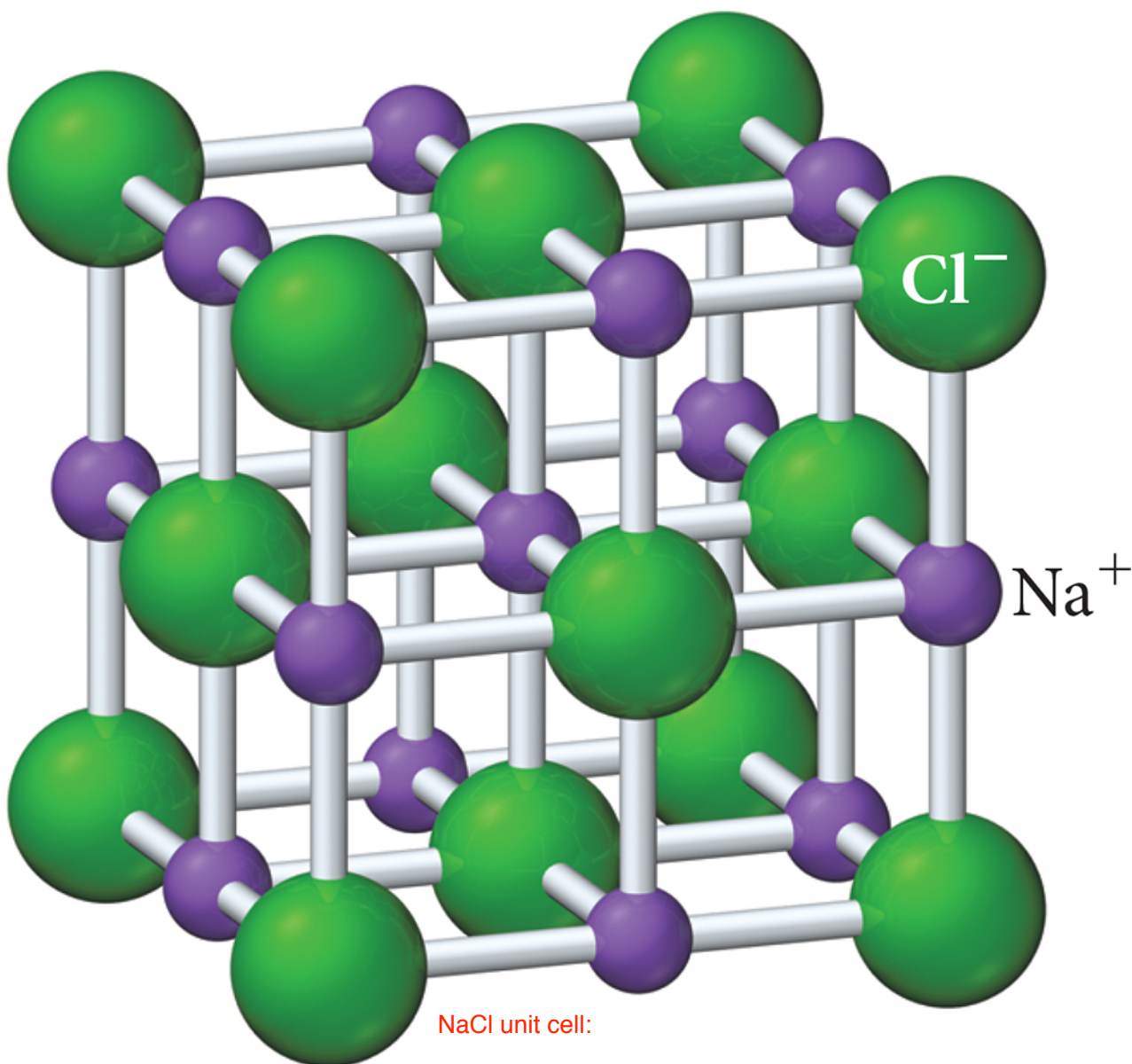
$$1 \text{ mol Fe atoms} = 55.85 \text{ g} = 6.022 \times 10^{23} \text{ atoms Fe}$$

$$1 \text{ atom Fe} = \frac{55.85 \text{ g}}{6.022 \times 10^{23}} = 9.27433 \times 10^{-23} \text{ g}$$

$$4 \text{ atoms Fe} = 4 \times M_{\text{Fe}} = 3.70973 \times 10^{-22} \text{ g}$$

$$d = \frac{m}{V} = 2.304 \text{ g/cm}^3$$

Sodium chloride (NaCl)

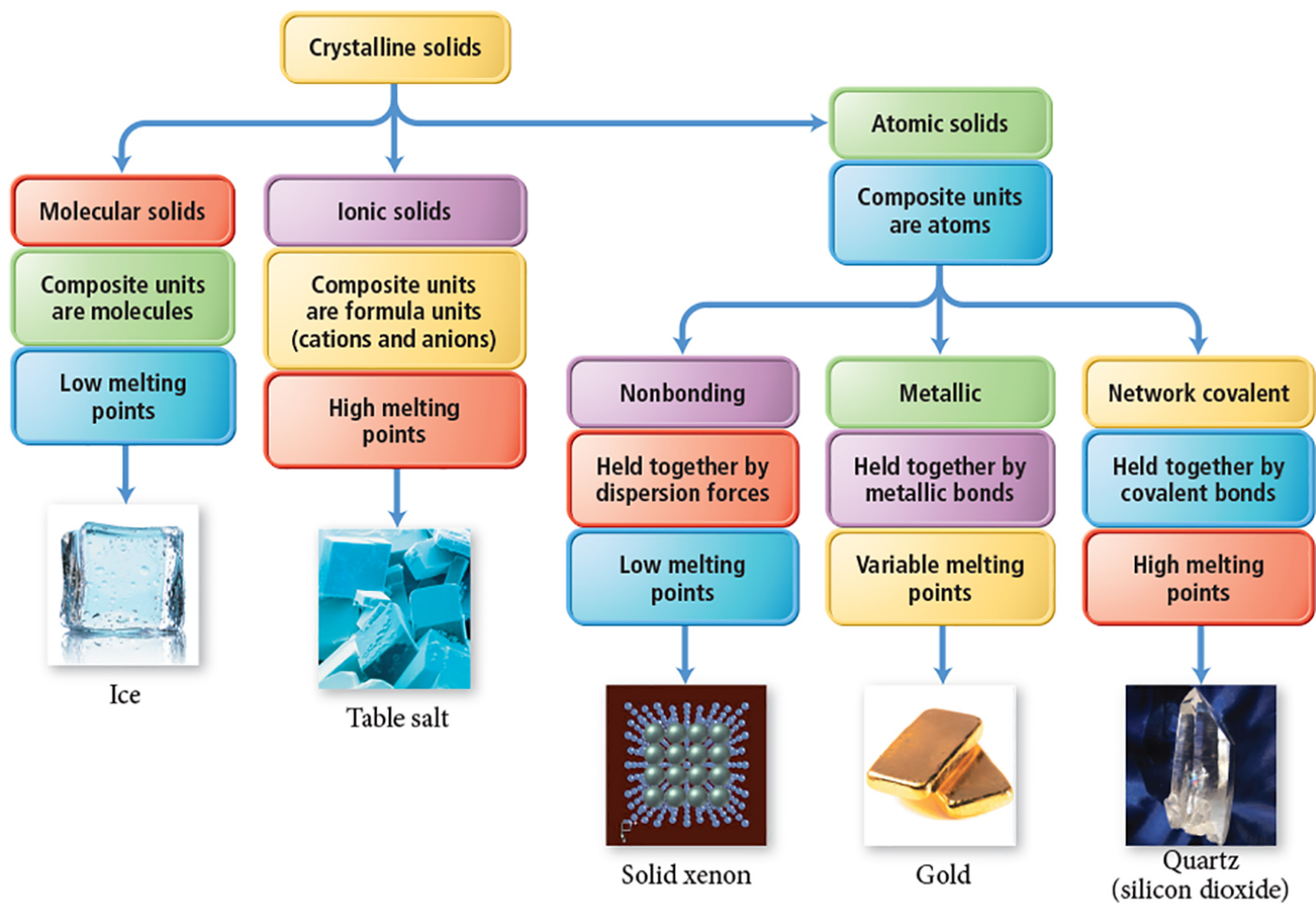


NaCl unit cell:

8 chlorine ions @ corners = $8 \times \frac{1}{8} = 1$
6 chlorine ions @ edges = $6 \times \frac{1}{2} = 3$
• Total chlorine ions = 4 / unit cell

12 sodium ions @ edges = $12 \times \frac{1}{4} = 3$
1 sodium ion @ center = $1 \times 1 = 1$
• Total sodium ions = 4 / unit cell

Formula: 4 Na ions : 4 Cl ions = 1:1 = NaCl

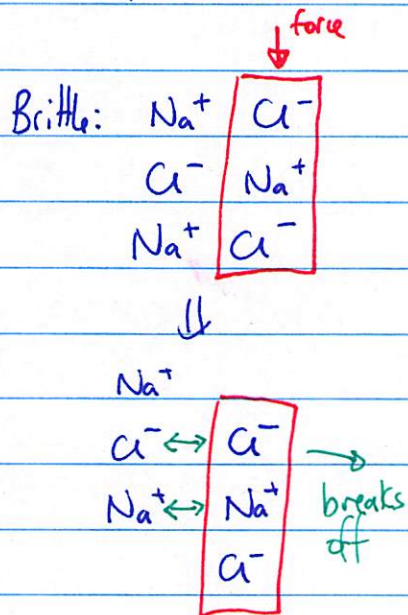


Kinds of crystalline solids

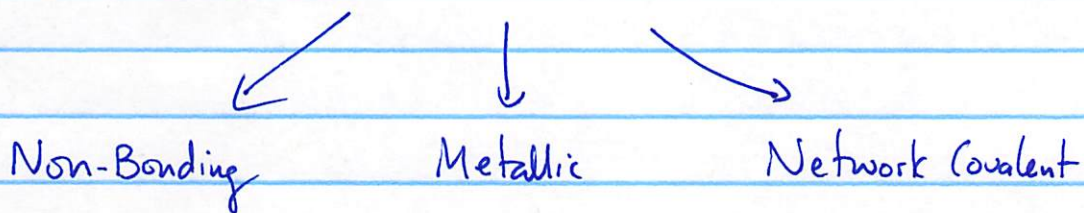
Molecular	Ionic	Atomic (Solids)	} Building blocks } held together by... } properties
IMFs (LDF, d-d, H-Bonds)	ion-ion (ionic bond)	:	
low mp. CH ₄ (s) -180°C H ₂ O(s) 0°C Sucrose(s) ~200°C C ₁₂ H ₂₂ O ₁₁ (s)	v. high mp/brittle ex: NaCl(s), 800°C Al ₂ O ₃ (s), 2000°C	greatest attraction	

- molecules on lattice points in unit cells

- cations/anions on lattice points in unit cells



Atomic Solids



LDF

Metallic bonds.

- e⁻ sea Ch 9.11

} held together by

low mp.

moderate → high mp

mp

} properties

ex: Xe(s) -110°C

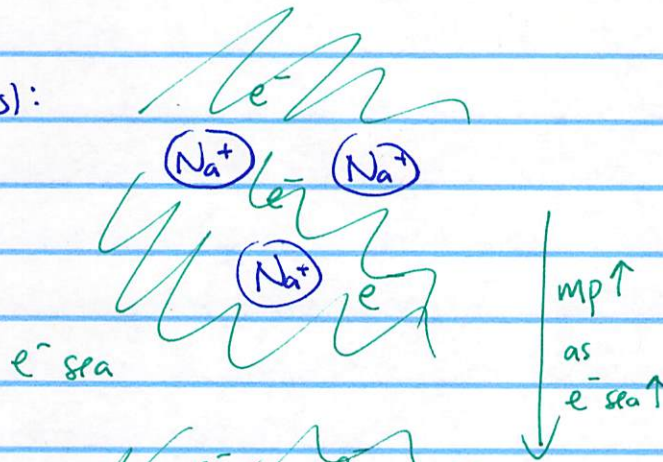
ex: Na(s) 100°C

He(s) -272°C

Mg(s) 650°C

Al(s) 660°C

Na(s):



Al(s)

